

Course Code: B20HS1101						
SAGI RAMA KRISHNAM RAJU ENGINEERING COLLEGE (A)				R 20		
I B. Tech I Semester –MODEL PAPER						
ENGLISH						
Common to AIDS,CE,CSE,ECE,EEE,IT&ME						
Time : 3Hrs.			Max. Marks: 70			
Answer any one Question from EACH UNIT.						
All questions carry equal marks.						
				M	CO	KL
UNIT-I						
<b>1</b>	<b>a.</b>	Compose an essay on “A Drawer full of happiness”.	7	2	K3	
	<b>b.</b>	Identify & Write suitable verb forms for the following. i) He _____ (work) in the college when his brother studied his engineering. ii) He _____ (finish) his task before his friend visited him. iii) As soon as he _____ (get) the telegram, at once he started. iv) It is high time she _____ (do) her project.. v) The book _____ (comprise) five chapters. vi) The photo of my grandfather _____ (hang) on the wall. vii) Neither team _____ (score) any goal.	7	5	K3	
OR						
<b>2</b>	<b>a.</b>	Compose an essay on any ONE of the following. i) Pros and cons of Social Networking Sites ii) The essence of education	7	2	K3	
	<b>b.</b>	Read the following passage and answer the questions that follow:  Liquids are practically incompressible. Unlike gases but like solids, a liquid does not change much in volume when the pressure on it is changed, even when the pressures of thousands of atmospheres are involved. The kinetic theory accounts for this saying that the amount of free space between the molecules of a liquid has been reduced almost to a minimum. Any attempt to compress the liquid meets with resistance as the electron cloud of one molecule repels the electron cloud of the adjacent molecule. Liquids diffuse slowly, but in gases it is more rapid. It occurs because molecules have kinetic energy and move from one place to another .In a liquid ,molecules do not move very far before they collide with neighbouring molecules. i) What is the nature of the liquids? ii) What does kinetic theory say about the incompressible nature of liquids? iii) What are the different kinds of matter? iv) Give the meaning for ‘diffusion’? v) Why diffusion is more rapid in gases? vi) Give the antonym for ‘kinetic’. vii) Mention a suitable title.	7	2,5	K3	

<b>UNIT-II</b>					
<b>3</b>	<b>a.</b>	Compose an essay on the relevance of Nehru's letter to Indira to the present context.	7	2	K3
	<b>b.</b>	Write antonyms and construct sentences for any SEVEN of the following words . i) procreate ii) hectic iii) reckon iv) beguile v) opulent vi) suffuse vii) astute viii) mandatory	7	5	K3
<b>OR</b>					
<b>4</b>	<b>a.</b>	Formulate a precis for the following passage. Most of us use the products of science – railways, aeroplanes, electricity, wireless and thousands of others – without thinking how they came into existence. We take them for granted, as if we were entitled to them as a matter of right. We are very proud of the fact that we live in an advanced age and are ourselves very advanced. Now, there is no doubt that our age is very different from previous ages and I think it is perfectly correct to say that is far more advanced. But it is a different thing from saying that we, as individuals or groups, are more advanced. It would be the height of absurdity to say that because an engine driver can run an engine and Plato or Socrates could not, the engine driver is more advanced than, or is superior to, Plato or Socrates. But it would be perfectly correct to say that the engine itself is a more advanced method of locomotion than Plato's chariot was.	7	2	K3
	<b>b.</b>	Write meanings and bring out the difference in usage for any FOUR of the following homonyms. i) Affect/Effect ii) Continuous/ Continual iii) Veracious/Voracious iv) Facilitate/Felicitate iv) Apposite/Opposite v) Intelligent/ Intelligible	7	5	K3
<b>UNIT-III</b>					
<b>5</b>	<b>a.</b>	Compose an essay on "Stephen Hawking – Positivity 'Benchmark'".	7	2	K3
	<b>b.</b>	Draft a letter to the educational consultancy asking about the information regarding the post-graduation and research programmes in foreign universities.	7	4	K3
<b>OR</b>					
<b>6</b>	<b>a.</b>	Write an E-mail to the manufacturer complaining about the computer that you bought recently.	7	4	K3
	<b>b.</b>	Draft a resume and covering letter for the post of a software engineer.	7	4	K3
<b>UNIT-IV</b>					
<b>7</b>	<b>a.</b>	Summarize the essay "Liking a Tree, Unbowed"?	7	2	K3
	<b>b.</b>	Give the meaning and write sentences of any SEVEN of the following FOREIGN PHRASES. i) ab initio ii) a la mode iii) siesta iv) amour proper v) ad hoc vi) Alma Mater vii) alter ego viii) bonafide	7	5	K3
<b>OR</b>					
<b>8</b>	<b>a.</b>	Write a pamphlet on book exhibition/ tour.	7	2	K3
	<b>b.</b>	Construct Sentences through the transformation of the following sentences as indicated.	7	5	K3

		<p>i. They painted the house green. (Add a question tag)</p> <p>ii. Do it. (Change the voice)</p> <p>iii) He said to the boy, “What are you doing here.” (Convert it into Indirect Speech)</p> <p>iv) My father said that Honesty is the best policy. (Convert it into direct Speech)</p> <p>v) They were very poor. They had insufficient food for themselves. They invited the strangers to dinner. (Change it into Complex)</p> <p>vi) The old man being weak could not walk properly. (Change it into Compound)</p> <p>vii). Shakespeare is the most famous of all writers in English. (Change it into positive degree)</p>			
<b>UNIT-V</b>					
<b>9</b>	<b>a.</b>	Elaborate the message which the author communicates to the readers through the lesson “Stay Hungry-Stay Foolish”.	7	2	K3
	<b>b.</b>	<p>Correct and Re-construct any SEVEN of the following Sentences.</p> <p>i) One must use his best efforts if he wishes to succeed.</p> <p>ii) Since he came, we are happy.</p> <p>iii) I could hardly believe in my eyes.</p> <p>iv) Suppose, if you arrive late, you will miss the show.</p> <p>v) Neither Jack is intelligent nor hardworking.</p> <p>vi) Hardly the sun had risen when we set out.</p> <p>vii) It is high time she improves her behaviour.</p> <p>viii) She gave me many informations.</p>	7	5	K3
<b>OR</b>					
<b>10</b>	<b>a.</b>	Compile a report to the editor about the problem of brain drain in India	7	4	K3
	<b>b.</b>	<p>Fill in the blanks with the appropriate choices.</p> <p>i) The film Titanic was promoted with all the usual _____</p> <p>a) hyperbole    b) dialect    c) taboo    d) aesthetic</p> <p>ii) The schedule of a few planes was _____ due to heavy smog.</p> <p>a) prohibited    b) abated    c) impeded    d) bolstered</p> <p>iii) Einstein had never bothered by the flood of _____ from his fellow critics.</p> <p>a) recantation    b) castigation    c) vituperation    d) skepticism</p> <p>iv) The field had been _____ by heavy downpour last night.</p> <p>a) tirade    b) fluctuated    c) mixed    d) saturated</p> <p>v) Modi is good at giving _____ speeches.</p> <p>a) extempore    b) prepared    c) epilogue    d) long</p> <p>vi) The manuscript was reproduced in _____ .</p> <p>a) facsimile    b) archives    c) cache    d) vacillation</p> <p>vii) Examine the report carefully before you _____ it publicly in front of the press and media.</p> <p>a) rescind    b) repudiate    c) revere    d) redress</p>	7	5	K3

\*\*\*\*\*

Course Code: B20BS1101				
SAGI RAMA KRISHNAM RAJU ENGINEERING COLLEGE (A)			R20	
I B.Tech I Semester –MODEL PAPER				
MATHEMATICS – I				
(LINEAR ALGEBRA AND DIFFERENTIAL EQUATIONS)				
Common to AIDS, CE, CSE, ECE, EEE, IT & ME				
TIME: 3 Hrs.		Max. Marks: 70 M		
Answer any One Question from Each Unit				
All questions carry equal marks				
UNIT-I		M	CO	KL
1.a)	Solve the system of equations $20x + y - 2z = 17$ , $3x + 20y - z = -18$ , $2x - 3y + 20z = 25$ by Gauss –Siedel method.	7	CO1	K3
b)	Investigate the values of $\lambda$ and $\mu$ so that the equations $2x + 3y + 5z = 9$ ; $7x + 3y - 2z = 8$ ; $2x + 3y + \lambda z = \mu$ ; have (i) no solution (ii) unique solution (iii) infinite number of solutions.	7	CO1	K3
(OR)				
2. a)	Solve the system of equations $10x + y+z =12$ , $2x+10y+z =13$ , $2x+2y+10z =14$ by Gauss- elimination method.	7	CO1	K3
b)	Define rank and determine the rank of the matrix A by reducing it to its normal form where A is: $A = \begin{bmatrix} 2 & 3 & -1 & -1 \\ 1 & -1 & -2 & -4 \\ 3 & 1 & 3 & -2 \\ 6 & 3 & 0 & -7 \end{bmatrix}$	7	CO1	K3
UNIT-II				
3.a)	Verify Cayley-Hamilton theorem and use the theorem to determine the inverse of the matrix $A = \begin{bmatrix} 1 & 0 & 3 \\ 2 & 1 & -1 \\ 1 & -1 & 1 \end{bmatrix}$	7	CO2	K3
b)	Reduce the quadratic form $2x^2 + 2y^2 + 2z^2 - 2xy - 2yz - 2zx$ to canonical form by the use of an orthogonal transformation.	7	CO2	K3
(OR)				
4. a)	Determine the Eigen values and the corresponding Eigen vectors of the matrix $A = \begin{bmatrix} 8 & -6 & 2 \\ -6 & 7 & -4 \\ 2 & -4 & 3 \end{bmatrix}$	7	CO2	K3
b)	If $A = \begin{bmatrix} 3 & 1 \\ -1 & 2 \end{bmatrix}$ , use Cayley-Hamilton theorem to determine the value of $2A^5 - 3A^4 + A^2 - 4I$ . Also determine the inverse of A.	7	CO2	K3

<b>UNIT-III</b>				
5.a)	Solve $\frac{dy}{dx} + (\tan x)y = (\sec x)y^3$ .	7	CO3	K3
b)	Solve $(y^4 + 2y)dx + (xy^3 + 2y^4 - 4x)dy = 0$ .	7	CO3	K3
<b>OR</b>				
6. a)	Determine the orthogonal trajectories of the family of parabolas $ay^2 = x^3$ .	7	CO4	K3
b)	A body originally at $80^\circ\text{C}$ , cools down to $60^\circ\text{C}$ in 20 minutes, the temperature of air being $40^\circ\text{C}$ . Determine the temperature of the body after 40 minutes from the original?	7	CO4	K3
<b>UNIT-IV</b>				
7.a)	Solve $(D^3 - D)y = 2x + 1 + 4 \cos x$ .	7	CO5	K3
b)	Solve $\frac{d^2y}{dx^2} - 2\frac{dy}{dx} + y = e^x \log x$ by the method of variation of parameters.	7	CO5	K3
<b>OR</b>				
8. a)	Solve $(D^2 + 3D + 2)y = e^{e^x}$ .	7	CO5	K3
b)	Use method of variation of parameters to solve the equation $\frac{d^2y}{dx^2} + y = \operatorname{cosec} x$ .	7	CO5	K3
<b>UNIT-V</b>				
9.a)	Determine $L\{t \cos at\}$ and $L\left\{\int_0^t e^{-t} \cos t dt\right\}$ .	7	CO6	K3
b)	Using convolution theorem evaluate $L^{-1}\left\{\frac{1}{(s+a)(s+b)}\right\}$ .	7	CO6	K3
<b>OR</b>				
10. a)	Determine $L^{-1}\left\{\frac{5s+3}{(s-1)(s^2+2s+5)}\right\}$ .	7	CO6	K3
b)	Solve $\frac{d^2y}{dt^2} + 4\frac{dy}{dt} + 3y = e^{-t}$ , $y(0) = y'(0) = 1$ by using Laplace transforms.	7	CO6	K3

\*\*\*\*\*

## I B. Tech I Semester - MODEL QUESTION PAPER

## APPLIED PHYSICS

(Common to AIDS, CE, EEE &amp; ME)

Time: 3 Hrs.

Max. Marks: 70

Answer any one Question from Each Unit

All questions carry equal Marks

M CO KL

## UNIT-I

1	a.	Explain the principle of Superposition and discuss the conditions for maxima and minima of intensity.	6	1	2
	b.	How the Newton's Rings are formed and deduce an expression for the wave length of light used.	8	1	3

## OR

2	a.	Distinguish the different types of diffractions of light.	6	1	2
	b.	Give qualitative and quantitative analysis of Fraunhofer diffraction at a single slit	8	1	3

## UNIT-II

3	a.	Define polarization and explain the different types of polarization possible in a dielectric	6	2	2
	b.	Deduce the Clausius Mosottirelation and its significance in dielectrics.	8	2	3

## OR

4	a.	Define Magnetic susceptibility and give a classification of magnetic materials.	6	2	1
	b.	Describe the Hysteresis exhibited by Ferromagnetic materials and explain its using a suitable theory	8	2	3

## UNIT-III

5	a.	Give the selection procedure of the active medium of laser device.	6	3	2
	b.	With suitable diagrams, discuss the working principle, design and working of He – Ne laser system	8	3	3

## OR

6	a.	What is the significance of Numerical Aperture of an optical fiber and obtain an expression for it.	8	3	2
	b.	Discuss the sensor applications of optical fibers.	6	3	3

## UNIT-IV

7	a.	What is an intrinsic semiconductor and obtain an expression for the density of carriers.	8	4	3
	b.	Distinguish between direct and indirect band gap semiconductors and mention their applications.	6	4	2

## OR

8	a.	Discuss the Hall Effect in detail and explain its significance.	8	4	3
	b.	Distinguish between drift and diffusion current in semiconductors.	6	4	2

<b>UNIT-V</b>					
9	a.	Explaining Magnetostriction effect, describe how the ultrasonics can be produced.	8	5	3
	b.	Mention the applications of ultrasonics.	6	5	1
<b>OR</b>					
10	a.	How the nano materials can be produced by sol – gel method.	6.	5	3
	b.	Write about Carbon Nanotubes	4	5	2
	C.	Discuss some important applications of nanomaterials.	4	5	1

\*\*\*\*\*

**ENGINEERING DRAWING**

(Common to CE,EEE &amp; ME)

**Time: 3 Hrs.****Max. Marks:70****Answer any one Question from Each Unit****All questions carry equal Marks****UNIT-I****M****CO****KL**

1	An inelastic string 145 mm long has its one end attached to the circumference of a circular disc of 40 mm diameter. Draw the curve traced out by the other end of the string, when it is completely wound around the disc, keeping the string always tight.	14	1	K3
---	---	----	---	----

**OR**

2	Construct a parabola when the distance of the focus from directrix is 50mm. Also draw the tangent and normal at any point on the curve.	14	1	K3
---	---	----	---	----

**UNIT-II**

3	a. Draw the projections of the following points on the same ground line, keeping the projectors 25mm apart. (i) Point A in the HP and lying 20mm behind the VP; (ii) Point B is 40mm above the HP and 25mm in front of the VP; (iii) Point C is 25mm below the HP and 25mm behind the VP; (iv) Point D is 15mm above the HP and 50mm behind the VP.	7	2	K3
	b. Draw the projections of a 75mm long straight line in the following positions: (i) parallel to and 30mm above the HP and in the VP; (ii) perpendicular to the VP, 25mm above the HP and its one end in the VP; (iii) Inclined at $30^\circ$ to the HP and its one end 20mm above it, parallel to and 30mm in front of the VP.	7	2	K3

**OR**

4	A line AB, of 80 mm long has its end A, 15 mm in front of VP and 20 mm above HP. The other end B is 40 mm above HP and 50 mm in front of VP. Draw the projections of the line and determine the inclinations of the line with HP and VP.	14	2	K3
---	--	----	---	----

**UNIT-III**

5	Draw a rhombus of diagonals 100 mm and 60 mm long, with the longer diagonal horizontal. The figure is the top view of a square of 100mm long diagonals, with a corner on the ground. Draw its front view and Determine the angle which its surface makes with the ground.	14	3	K3
---	---	----	---	----

**OR**



6	A semicircular plate of 40mm diameter has its straight edge in the VP and inclined at $45^{\circ}$ to the HP, the surface of the plate makes an angle of $30^{\circ}$ with the VP. Draw its projections.	14	3	K3
<b>UNIT-IV</b>				
7	A hexagonal pyramid, base 25mm side and axis 50mm long, has an edge of its base on the ground. Its axis is inclined at $30^{\circ}$ to the ground and parallel to the VP. Draw its projections.	14	4	K3
<b>OR</b>				
8	A hexagonal pyramid, base 30 mm side and axis 75 mm long, resting on its base on HP with two of its edges parallel to VP is cut by two section planes both perpendicular VP. The horizontal section plane cuts the axis at a point 35 mm from the apex. The other plane which makes an angle of $45^{\circ}$ with the HP also intersects the axis at the same point. Draw the front view, sectional top view and true shape of section	14	4	K3
<b>UNIT-V</b>				
9	A cone of base diameter 50 mm and axis 60 mm is resting on its base on the HP. Draw the development of its lateral surface when it is cut by an auxiliary inclined plane inclined at $60^{\circ}$ to the HP and bisecting the axis.	14	5	K3
<b>OR</b>				
10	A square pyramid has its base side 40mm and height 60mm. Draw the isometric projection of the pyramid when the axis of the solid is horizontal and vertical.	14	5	K3

\*\*\*\*\*

Course Code: B20 CE1101					
SAGI RAMA KRISHNAM RAJU ENGINEERING COLLEGE (A)					R 20
I B. Tech I Semester - MODEL QUESTION PAPER					
ENGINEERING GEOLOGY					
Civil Engineering					
Answer any one Question from Each Unit					
All questions carry equal Marks					
			M	CO	KL
<b>UNIT - I</b>					
1.	a).	Define Engineering Geology. Describe about branches of geology and its importance in Civil Engineering works.	7	1	2
	b).	Define Weathering and explain the types of Weathering. Mention the civil engineering importance of weathering.	7	2	2
<b>OR</b>					
2.	a).	Describe the landforms produced by running water and describe about its civil engineering importance.	7	2	2
	b).	Explain the importance of Geology in Civil Engineering with case studies.	7	1	2
<b>UNIT - II</b>					
3.	a).	What is mineral? How do you identify minerals with help of physical properties of minerals.	7	3	2
	b).	Explain Feldspars and quartz group minerals	7	3	2
<b>OR</b>					
4.	a).	What are igneous rocks? Write about textures, structures and forms of Igneous rocks and write its engineering importance.	7	3	2
	b).	Explain Granite, Sandstones and Quartzite.	7	3	2
<b>UNIT - III</b>					
5.	a).	Define faults and joints and explain their significance in civil engineering.	7	4	2
	b).	Explain the terms-Strike, Dip, Outcrop	7	4	2
<b>OR</b>					
6.	a).	Define the term fold? Explain parts of the fold with neat sketch.	7	4	2
	b).	Describe about classification of folds with neat sketches and their importance in Civil Engineering.	7	4	3
<b>UNIT - IV</b>					
7.	a).	Describe Engineering properties of rocks.	7	3	3
	b).	Explain the geological control of groundwater movement.	7	4	3
<b>OR</b>					
8.	a).	How do you identify soil profile using electrical Resistivity method?	7	4	3
	b).	Explain causes, effects and control measures of landslides.	7	3	2
<b>UNIT - V</b>					
9.		Evaluate Geological Investigations for dams and reservoirs	14	5	4
<b>OR</b>					
10.	a).	Define tunnel and explain the purpose of tunnelling.	7	5	2
	b).	Influence of geology for successful tunnelling.	7	5	3

UNIT-I		M	CO	KL
1.a)	Determine the Fourier series for the function $f(t) = \begin{cases} -1, & -\pi < t < -\pi/2 \\ 0, & -\pi/2 < t < \pi/2 \\ 1, & \pi/2 < t < \pi \end{cases}$	7	CO1	K3
b)	Determine Fourier series of the function $f(x) = 2x - x^2$ in $(0, 3)$ and hence deduce that $\frac{1}{1^2} - \frac{1}{2^2} + \frac{1}{3^2} - \frac{1}{4^2} + \dots = \frac{\pi}{12}$ .	7	CO1	K3
<b>OR</b>				
2. a)	Determine Fourier series for the function $f(x)$ given by $f(x) = \begin{cases} 1 + \frac{2x}{\pi}, & \text{if } -\pi \leq x \leq 0 \\ 1 - \frac{2x}{\pi}, & \text{if } 0 \leq x \leq \pi \end{cases}$ and deduce that $\frac{1}{1^2} + \frac{1}{3^2} + \frac{1}{5^2} + \dots = \frac{\pi^2}{8}$ .	7	CO1	K3
b)	Determine the Half – Range cosine series for the function $f(x) = x^2$ in the range $0 \leq x \leq \pi$ .	7	CO1	K3
<b>UNIT-II</b>				
3.a)	Using the Fourier Sine Transform of $e^{-ax}$ ( $a > 0$ ), evaluate $\int_0^{\infty} \frac{x \sin kx}{a^2 + x^2} dx$ .	7	CO2	K3
b)	Using Fourier integral representation, establish that $\int_0^{\infty} \frac{\omega \sin \omega x}{1 + \omega^2} d\omega = \frac{\pi}{2} e^{-x}, x > 0$ .	7	CO2	K3
<b>OR</b>				
4. a)	Determine the inverse Fourier sine transform $f(x)$ of $F_s(p) = \frac{p}{1+p^2}$ .	7	CO2	K3
b)	Using Parseval's Identity, establish that $\int_0^{\infty} \frac{x^2}{(1+x^2)^2} dx = \frac{\pi}{4}$ .	7	CO2	K3
<b>UNIT-III</b>				
5.a)	If $U = \tan^{-1} \frac{x^3+y^3}{x-y}$ and $x U_x + y U_y = \sin 2U$ , establish that $x^2 U_{xx} + 2xy U_{xy} + y^2 U_{yy} = 2 \cos 3U \sin U$ .	7	CO3	K3
b)	If $u = x^2 - 2y^2, v = 2x^2 - y^2$ where $x = r \cos \theta, y = r \sin \theta$ then establish that $\frac{\partial(u,v)}{\partial(r,\theta)} = 6 r^3 \sin 2\theta$ .	7	CO3	K3
<b>OR</b>				

6. a)	Develop $x^2y + 3y - 2$ in a Taylor series in powers of $(x - 1)$ and $(y + 2)$ using Taylor's theorem.	7	CO4	K3
b)	By using the method of differentiation under the integral sign establish that $\int_0^\infty \frac{\tan^{-1}(ax)}{x(1+x^2)} dx = \frac{\pi}{2} \log(1+a)$ , $a \geq 0$ .	7	CO4	K3
<b>UNIT-IV</b>				
7. a)	Solve $x^2(y-z)p + y^2(z-x)q = z^2(x-y)$ .	7	CO5	K3
b)	solve $(D^2 - DD' - 2D'^2)z = (y-1)e^x$ .	7	CO5	K3
<b>OR</b>				
8. a)	Solve $x(y-z)p + y(z-x)q = z(x-y)$ .	7	CO5	K3
b)	solve $(D + D' - 1)(D + 2D' - 3)z = 3x + 6y + 4$ .	7	CO5	K3
<b>UNIT-V</b>				
9.a)	Determine the solution of $\frac{\partial u}{\partial x} + \frac{\partial u}{\partial y} = 0$ by the method of separation of variables.	7	CO6	K3
b)	A tightly stretched elastic string of length $L$ , fixed at its end points is initially in a position given by $u(x, 0) = u_0 \sin^3 \frac{\pi x}{L}$ . If it is released from rest, determine the displacement at any subsequent time.	7	CO6	K3
<b>OR</b>				
10.a)	Determine the solution of $x \frac{\partial u}{\partial x} + y \frac{\partial u}{\partial y} = 0$ by the method of separation of variables.	7	CO6	K3
b)	A bar of conducting material of length $\pi$ units is initially kept at a temperature $\sin x$ . Determine the temperature at any subsequent time if the ends of the bar are held at zero temperature.	7	CO6	K3

\*\*\*\*\*

## APPLIED CHEMISTRY

Common to AIDS,CE,EEE &amp; ME

Time: 3 Hrs.

Max. Marks:70

Answer any one Question from Each Unit

All questions carry equal Marks

			M	CO	KL
<b>UNIT-I</b>					
1	a.	Explain the mechanism of free radical Polymerization reaction with a suitable example.	7	CO1	K2
	b.	Distinguish between thermoplastic and thermosetting resins	7	CO1	K3
<b>OR</b>					
2	a.	What are conducting Polymers? Discuss the applications of conducting Polymers.	7	CO1	K2
	b.	Write notes on Bu Na – S and Bu Na – N.	7	CO1	K2
<b>UNIT-II</b>					
3	a.	Explain the Proximate analysis of coal and give its significance.	7	CO3	K2
	b.	Explain Photovoltaic cell with neat diagram.	7	CO3	K2
<b>OR</b>					
4	a.	Explain the fractional distillation of crude oil	7	CO3	K2
	b.	How synthetic Petrol can be prepared by Berguis Process.	7	CO3	K2
<b>UNIT-III</b>					
5	a.	Explain the mechanism of electrochemical theory of corrosion with neat diagram	7	CO2	K2
	b.	Describe briefly about cathodic Protection.	7	CO2	K3
<b>OR</b>					
6	a.	Explain Hydrogen – Oxygen fuel cell with neat cell diagram	7	CO4	K2
	b.	Discuss about Lithium batteries	7	CO4	K2
<b>UNIT-IV</b>					
7	a.	What is hardness? How it is determined by EDTA method? Explain	7	CO2	K2
	b.	Describe with equations how water can be softened using Lime & Soda Process	7	CO2	K2
<b>OR</b>					
8	a.	Discuss various sterilizing methods used in municipal water treatment	7	CO2	K2
	b.	Illustrate the reverse osmosis Process with a neat diagram	7	CO2	K2
<b>UNIT-V</b>					
9	a.	Discuss chemistry involved in setting and hardening of cement?	7	CO4	K2
	b.	What are refractories? Discuss the classification of refractories.	7	CO4	K2
<b>OR</b>					
10	a.	Write the engineering applications of Liquid Crystals.	7	CO3	K2
	b.	Explain the stoichiometric defects in crystals.	7	CO3	K2

\*\*\*\*\*

**Time: 3 Hrs.****Max. Marks:70**

Answer any one Question from Each Unit

All questions carry equal Marks

**M CO KL****UNIT-I**

1	a.	What do you mean C Variable, C Constants? Explain in detail	7M	CO1	K1
	b.	Explain Expressions Precedence and Associativity in detail	7M	CO1	K2

**OR**

2	a.	Explain types of storage classes in C	7M	CO1	K2
	b.	Write the basic structure of C program and explain each and every step	7M	CO1	K2

**UNIT-II**

3	a.	Explain logical, bitwise operators with examples	7M	CO2	K2
	b.	Differentiate two-way selection and multi-way selection	7M	CO2	K2

**OR**

4	a.	What is the difference between Pre-test loop and Post-test Loop with an example	7M	CO2	K2
	b.	Discuss Event and Counter controlled loops	7M	CO2	K2

**UNIT-III**

5	a.	What is an Array? Define and initialize 1-D, 2-D Arrays	7M	CO3	K1
	b.	Write a program to evaluate the average of values in an Array	7M	CO3	K3

**OR**

6	a.	Explain declaration and initialization of Array of Strings	7M	CO3	K2
	b.	What is an Union? List the difference between Structures and Unions	7M	CO3	K1

**UNIT-IV**

7	a.	Explain the concept of Array of Pointers with example	7M	CO4	K2
	b.	With proper example explain different arithmetic operators on Pointers	7M	CO4	K2

**OR**

8	a.	Explain L and R values in C Language and how are they used in C Language	7M	CO4	K2
	b.	Discuss various Processor Commands	7M	CO4	K2

**UNIT-V**

9	a.	How to pass Array to a Function in C	7M	CO5	K2
	b.	Discuss the Types of Functions in C	7M	CO5	K2

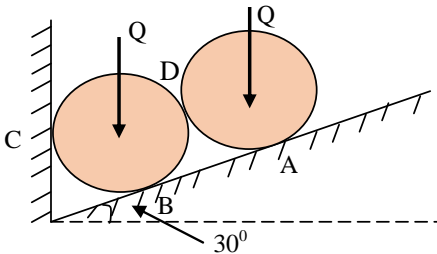
**OR**

10	a.	What is Stream? Describe two different methods of creating a Stream-Oriented data file	7M	CO5	K1
	b.	Write a program to Copy contents from one File to another File	7M	CO5	K1

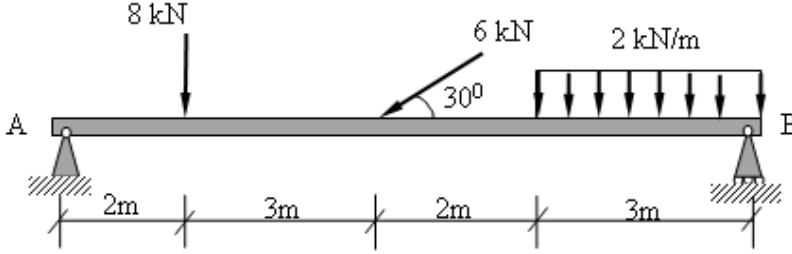
\*\*\*\*\*

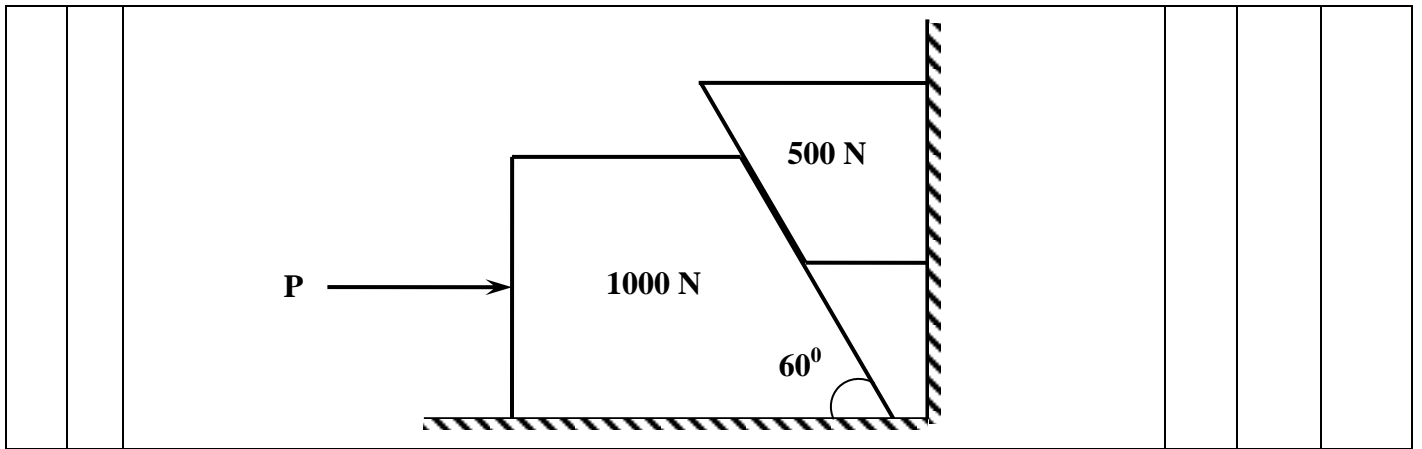
M CO KL

UNIT-I

1	a.	The following are the four forces acting at a point on a body. Find the resultant and its position. i. 500N at N 45° E ii. 100N at N 80° E iii. 300N at S 30° E iv. 600N at N 20° W	5	1	3
	b.	Two identical rollers, each of weight $Q = 100\text{ N}$ are supported by an inclined plane and a vertical wall as shown in Fig. Applying conditions of equilibrium, determine the reactions at points of support A, B and C. assume the surfaces are smooth. 	9	1	3

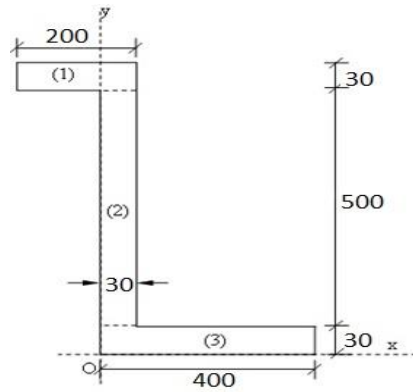
OR

2	a.	Determine the reactions at supports A and B for the loaded beam shown in Fig. 	5	1	3
	b.	Referring to the figure, the coefficients of friction are as follows: 0.25 at floor, 0.3 at the wall and 0.2 between the blocks. Determine the minimum value of a horizontal force 'P' applied to the lower block that will hold the system in equilibrium.	9	1	3



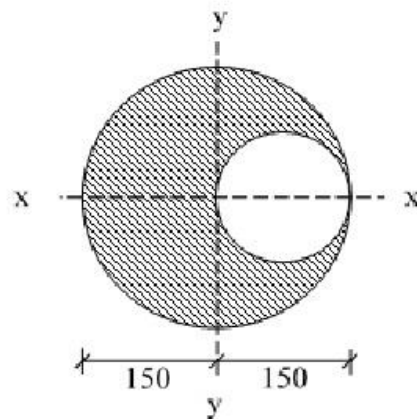
**UNIT-II**

3	a.	State and prove Parallel axis theorem.	7	2	3
	b.	Calculate the Centroid of Z – section shown in fig. (All units are in mm)	7	2	3



**OR**

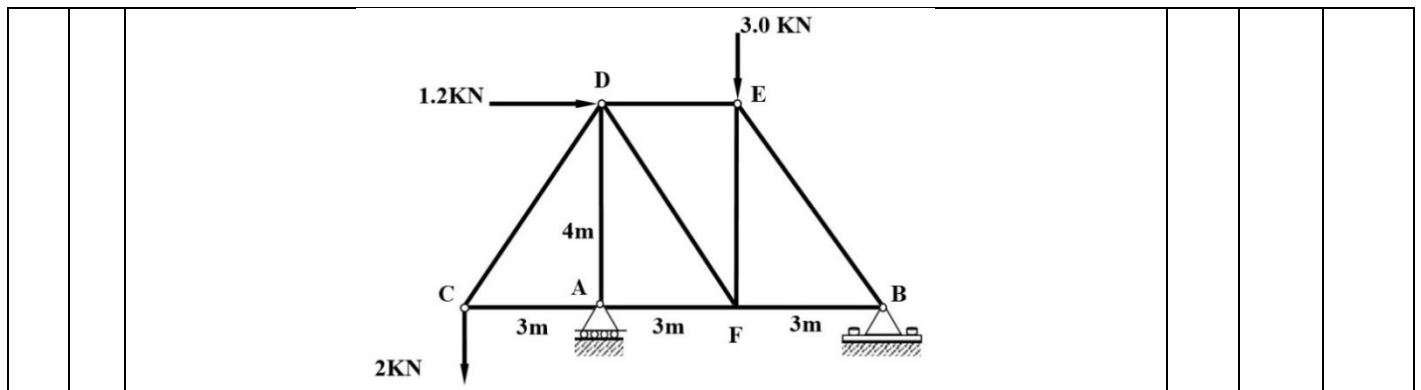
4		Determine moment of inertia for shaded region shown in fig.(All units are in mm)	14	2	3
---	--	--	----	---	---



**UNIT-III**

5		Solve for the Forces in all the members of the truss supported on roller at A and hinge at B.	14	3	3
---	--	---	----	---	---





**OR**

6		Determine the forces in all the members of the truss shown in the Fig.	<b>14</b>	<b>3</b>	<b>3</b>

**UNIT-IV**

7	a.	A particle moves along a straight line. Its motion is represented by the equation $S = 16t + 4t^2 - 3t^3$ where, S is in meters and t is in Seconds, Determine a. Displacement, velocity and acceleration 2 seconds after start: b. Displacement and acceleration when velocity is zero and c. Displacement and velocity when acceleration is zero	<b>7</b>	<b>4</b>	<b>3</b>
	b.	A stone dropped into a well is heard to strike the water after 4 seconds. Find the depth of the well, if the velocity of sound is 350 m/sec.	<b>7</b>	<b>4</b>	<b>3</b>

**OR**

8	a.	Find the height of the tower from top of which an object falls freely and during the last second of its motion, the object travels a distance is equal to $\frac{2}{3}$ of height of the tower. Take $g = 9.81 \text{ m/sec}^2$	<b>7</b>	<b>4</b>	<b>3</b>
	b.	A stone is dropped from the top of a tower. When it has fallen a distance of 10 m, another stone is dropped from a point 38 m below the top of the tower. If both the stones reach the ground at the same time, calculate (i) the height of the tower and (ii) the velocity of the stones when they reach the ground.	<b>7</b>	<b>4</b>	<b>3</b>

**UNIT-V**

9	a.	Find the tension S in the string during the motion of the system shown in fig. If $W_1 = 200 \text{ N}$ and $W_2 = 100 \text{ N}$ , the system is in the vertical plane and coefficient of friction between the inclined plane and block $W_1$ is $\mu = 0.2$ . Assume the pulley is without weight and friction.	<b>7</b>	<b>5</b>	<b>3</b>
---	----	---	----------	----------	----------

	b.	Two bodies, one of which is 400 N with a velocity of 8 m/sec and the other of 250 N with a velocity of 12 m/sec, move towards each other along a straight line and collide centrally. Find the velocity of each body after impact if the coefficient of restitution is 0.8. Also find the loss in Kinetic Energy?	7	5	3
<b>OR</b>					
10	a.	Two weights $W_1 = 400\text{ N}$ and $W_2 = 100\text{ N}$ are connected by a string and move along a horizontal plane under the action of force $P = 200\text{ N}$ applied horizontally to the weight $W_1$ . The coefficient of friction between the weights and the plane is 0.25. Determine the acceleration of the weights and the tension in the string.	7	5	3
	b.	A 5 m long ladder shown in Fig. begins to slip down the wall at A at a rate of 1 m/sec. Find the location of the instantaneous centre of rotation of the ladder. Also, find the velocity of point B at the base of the ladder using instantaneous centre method.	7	5	3

\*\*\*\*\*

Answer any one Question from Each Unit

All questions carry equal Marks

M CO KL

## UNIT-I

1	a.	Explain briefly the various Classifications of stones?	7	CO1	K2
	b.	What is seasoning of timber? Explain the defects in timber?	7	CO1	K2

## OR

2	a.	What are the steps involved in the manufacturing process of a Brick? Explain the Burning process of bricks in detail?	7	CO1	K2
	b.	Classify tiles and explain roof tiles?	7	CO1	K2

## UNIT-II

3	a.	Define Hydration of cement and Explain Bogues compounds?	7	CO2	K2
	b.	State and explain the constituents of cement along with their percentages?	7	CO2	K2

## OR

4	a.	What is an admixture? What are the effects of accelerators and retarders on the properties of concrete?	7	CO2	K2
	b.	Explain the setting times of cement with a neat sketch?	7	CO2	K2

## UNIT-III

5	a.	What are the deleterious materials which may be found in aggregates? Explain?	7	CO2	K2
	b.	Explain the phenomena of "Bulking of Sand" in detail?	7	CO2	K2

## OR

6	a.	Explain the importance of size, shape and texture of aggregates in concrete?	7	CO2	K2
	b.	What is Alkali Aggregate reaction? Explain?	7	CO2	K2

## UNIT-IV

7	a.	Explain any two laboratory tests for measurement of workability?	7	CO3	K2
	b.	Explain the effect of time and temperature on Workability of concrete?	7	CO3	K2

## OR

8	a.	Design the concrete mix for $M_{25}$ grade of concrete for the following data using IS method <ul style="list-style-type: none"> <li>i. Compressive strength for 28 days = 25 N/mm<sup>2</sup></li> <li>ii. Standard deviation = 4 N/mm<sup>2</sup></li> <li>iii. Maximum size of Coarse Aggregates = 20mm</li> <li>iv. Type of exposure : Moderate</li> <li>v. Max. W/C ratio : 0.5</li> <li>vi. Specific gravity of Fine aggregate and Coarse aggregate = 2.7</li> <li>vii. Fine aggregate is confined to Zone: II</li> <li>viii. Coarse Aggregates: 20mm MSA (78%) &amp; 10mm MSA (22%)</li> </ul>	15	CO3	K3
---	----	---	----	-----	----

## UNIT-V

9	a.	Explain the factors which influence the strength of hardened concrete?	7	CO4	K2
	b.	Briefly explain the rebound hammer test and Ultrasonic pulse velocity test?	7	CO4	K2
<b>OR</b>					
10	a.	Discuss briefly about Water-cement Ratio and Gel-Space ratio.	7	CO4	K2
	b.	Explain the relation between compression & tensile strength of hardened concrete?	7	CO4	K2

\*\*\*\*\*